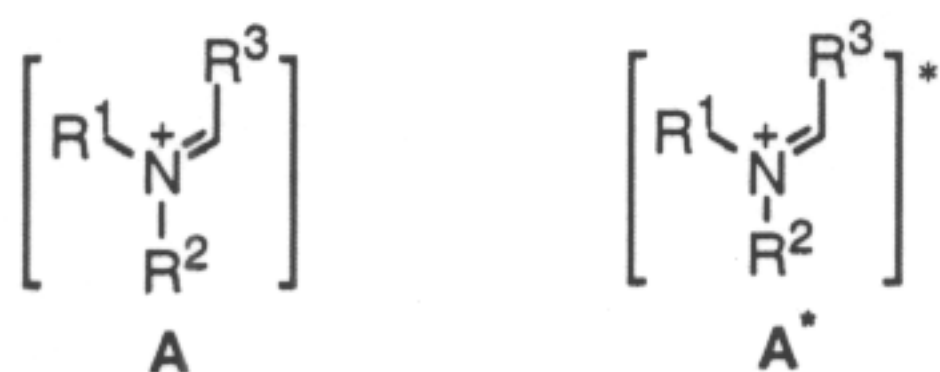


Electrochemical Generation of Optically Active Iminium Ion

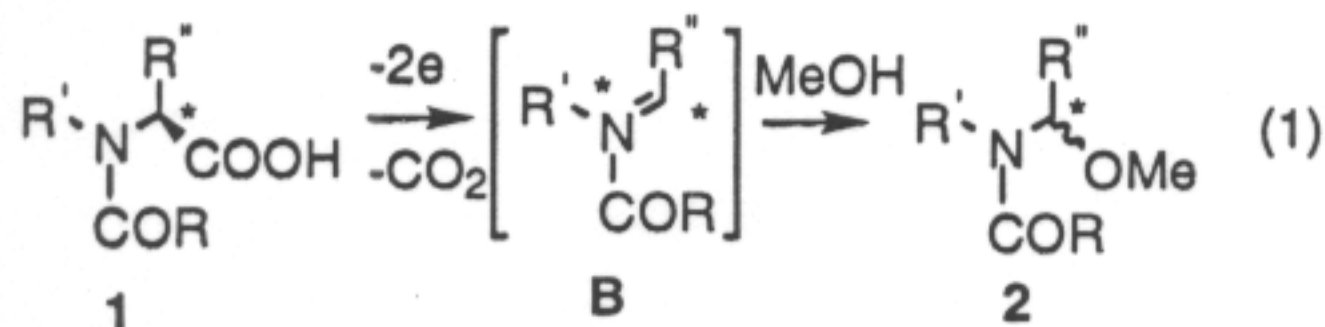
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Iminium ions **A** are typical carbenium ions stabilized by the adjacent nitrogen atom.

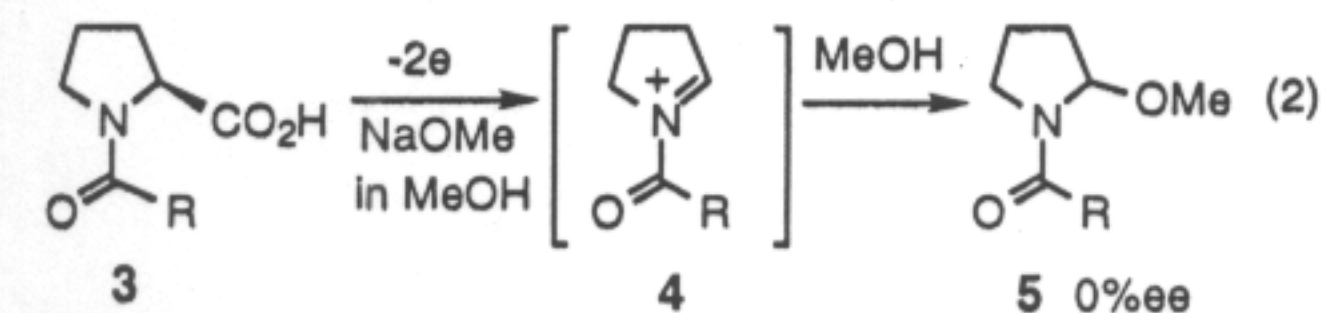


Thus, memory of chirality through iminium ions as intermediates has been recognized much difficult. Recently, however, we found a first memory of chirality through acyliminium ions **B** which are generated by electrochemical oxidation of *N*-acyl- α -amino acids **1** (eq 1).¹ We report herein further development of this finding as well as the application to the generation of optically active iminium ions **A**.



1) Memory of chirality through intermediates **B**

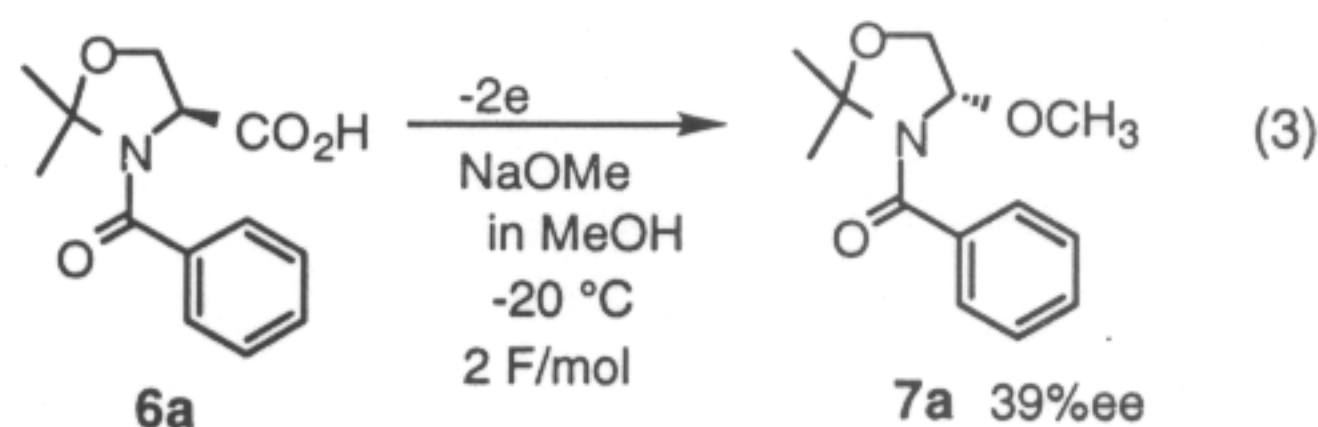
The non-Kolbe reaction of optically active *N*-acyl- α -amino acids **3** in methanol affords *N,O*-acetals **5**, in which the original chirality of **1** is completely lost because of an intervention of acyliminium ions **4** as intermediates (eq. 2).



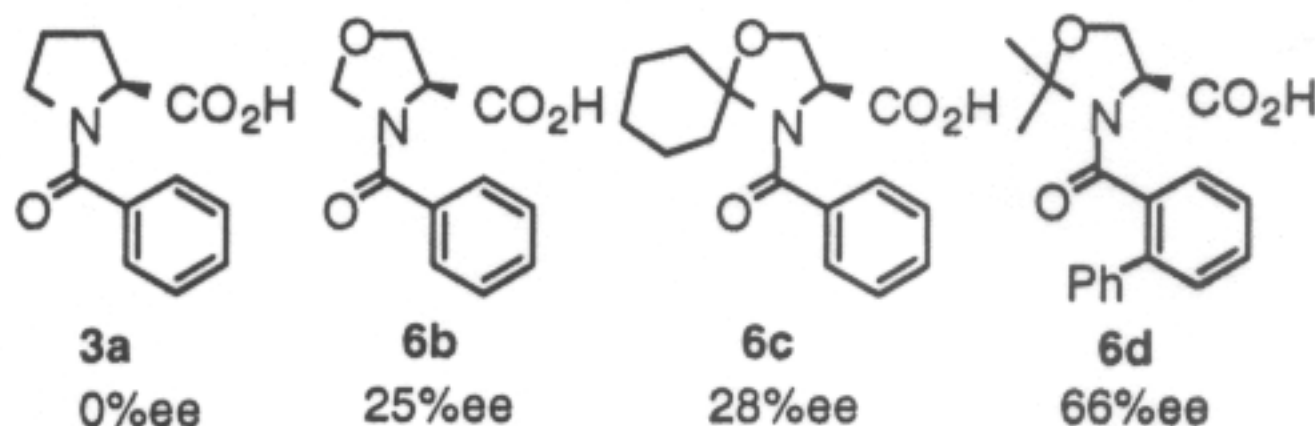
R=Ph, OMe

On the other hand, we found that the non-Kolbe electrolysis of **6** derived from L-serine gave the corresponding *N,O*-acetal **7** in 39%ee (eq. 3).¹

In order to improve the ee in the oxidation, we surveyed a variety of modified compounds **6b-d**.²

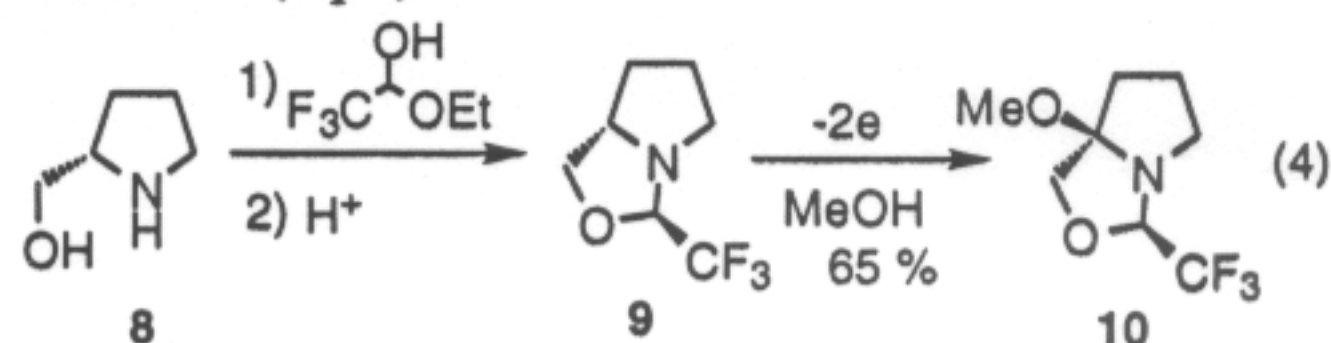


The result is as follows.



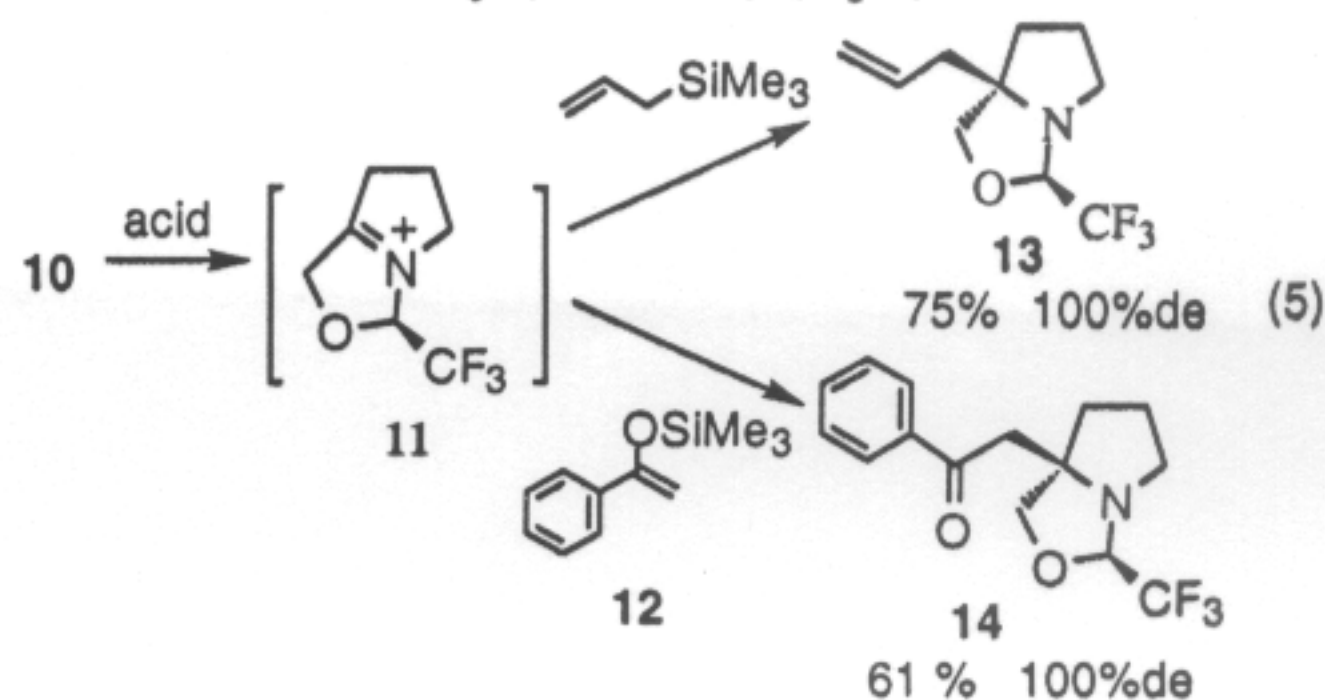
2) Memory of chirality through intermediates **A**³

N,O-acetal **10**, a precursor of iminium ion, was prepared by electrochemical oxidation of **9**, which was derived from optically active amino alcohol **8** (eq 4).



The oxidation potential (*E*_p) of **9** was 1.91 V vs Ag/AgCl. The value was largely positive than that of **8** (1.24 V), and more negative than that of *N*-methoxycarbonylpyrrolidine (2.17 V).

Treatment of **10** with CF₃CO₂H in CH₂Cl₂ gave iminium ion **11**, which could be trapped with carbon nucleophiles such as allyltrimethylsilane and silyl enol ether **12** to give the products **13** and **14**, respectively, with complete diastereoselectivity (100%de) (eq 5).



Iminium ion **11** was enough stable at rt to allow the measurement of its ¹H NMR spectrum; δ (CDCl₃) 6.35 for CF₃CH, 5.60 for OCH₂.

References

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- unpublished.